## 実験報告書様式(一般利用課題・成果公開利用)

( ※本報告書は英語で記述してください。ただし、産業利用課題として採択されている方は日本語で記述していただいても結構です。 )

| Experimental Report J-PARC  | 承認日 Date of Approval<br>承認者 Approver<br>提出日 Date of Report Nov. 5, 2017 |  |
|---|---|--|
| 課題番号 Project No.  | 装置責任者 Name of Instrument scientist                                      |  |
| 2016A0114   | Kaoru SHIBATA   |  |
| 実験課題名 Title of experiment   | 装置名 Name of Instrument/(BL No.)   |  |
| Direct observation of the mobile/immobile Li ions in                | DNA / BL 02   |  |
| Li <sub>7</sub> P <sub>3</sub> S <sub>11</sub> superionic conductor | 実施日 Date of Experiment  |  |
| 実験責任者名 Name of principal investigator                               | November 18-22, 2016  |  |
| Kazuhiro MORI   |   |  |
| 所属 Affiliation  |   |  |

試料、実験方法、利用の結果得られた主なデータ、考察、結論等を、記述して下さい。(適宜、図表添付のこと) Please report your samples, experimental method and results, discussion and conclusions. Please add figures and tables for better explanation.

| 1. 試料 | Name of sample(s) an | d chemical formula, | or compositions | including physical form. |
|-------|----------------------|---------------------|-----------------|--------------------------|
|-------|----------------------|---------------------|-----------------|--------------------------|

| 1. | $^{7}\text{Li}_{7}\text{P}_{3}\text{S}_{11}$ | crystal  |
|----|--|----------|
|    | <b>21/1</b> 3011                             | or , but |

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## 2. 実験方法及び結果(実験がうまくいかなかった場合、その理由を記述してください。)

Experimental method and results. If you failed to conduct experiment as planned, please describe reasons.

Li<sub>2</sub>S–P<sub>2</sub>S<sub>5</sub> glasses are promising candidates as solid electrolytes in all-solid-state lithium-ion batteries, because they show high ionic conductivities ( $\sim 10^{-4}$  S/cm) at room temperature (RT) [1]. Furthermore, Li<sub>7</sub>P<sub>3</sub>S<sub>11</sub> crystal, which can be prepared by aging (Li<sub>2</sub>S)<sub>70</sub>(P<sub>2</sub>S<sub>5</sub>)<sub>30</sub> glass, shows more than  $10^{-3}$  S/cm at RT [2]. The aim of this study is to observe the motion of Li ions in the  $^7$ Li<sub>7</sub>P<sub>3</sub>S<sub>11</sub> crystal. Quasi-elastic neutron scattering (QENS) is a powerful tool for directly observing the motion of ions in solids. Even though the incoherent scattering cross section for  $^7$ Li nucleus is relatively small ( $\sigma_{inc}(^7$ Li) = 0.78 barn), we can observe the QENS spectrum associated with the thermal self-diffusion for  $^7$ Li ions due to the high flax of neutron beam generated by the 1-MW pulsed-spallation neutron source in the J-PARC/MLF. In this work, we performed QENS experiments using the DNA at the BL02 beam line: the energy resolution of the DNA is 3.6  $\mu$ eV [3].

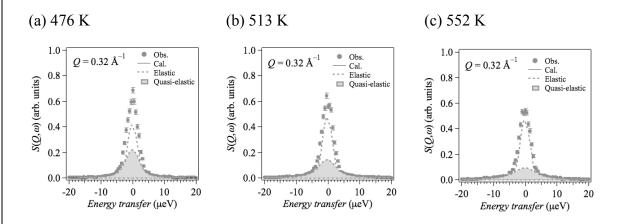
## 2. 実験方法及び結果(つづき) Experimental method and results (continued)

 $^{7}\text{Li}_{7}\text{P}_{3}\text{S}_{11}$  crystal was synthesized by mechanical alloying (MA). The three-dimensional structure and the predicted conduction pathways of Li ions in the  $^{7}\text{Li}_{7}\text{P}_{3}\text{S}_{11}$  crystal have been studied by combining the reverse Monte Carlo (RMC) modeling and the bond valence sum (BVS) approach using time-of-flight (TOF) neutron diffraction data (BL21 NOVA, J-PARC/MLF) [4]. The sample was placed in a cylindrical silica tube (12 mm in diameter and 0.5 mm in thickness), and then put into a cylindrical Al holder under a high-purity argon gas atmosphere. The QENS data were collected in the *Q* range of 0.07–1.86 Å $^{-1}$  at 150 K, 300 K, 476 K, 513 K, and 552 K, where *Q* is the magnitude of the scattering vector.

S(Q, E) data for the  ${}^{7}\text{Li}_{7}\text{P}_{3}\text{S}_{11}$  crystal were collected using the DNA. Figure 1 shows the temperature dependence of S(Q, E) spectrum for the  ${}^{7}\text{Li}_{7}\text{P}_{3}\text{S}_{11}$  crystal at  $Q = 0.32 \text{ Å}^{-1}$ . The QENS component around the elastic peak (E = 0) is clearly observed and becoming broad with increasing temperature. To get the dynamics information such as the jump distance, < l >, and the residence time,  $\tau_{0}$ , for Li ions, further investigations are now in progress.

## References

- [1] A. Hayashi et al., J. Am. Ceram. Soc. 84 (2001) 477.
- [2] F. Mizuno et al., Electrochem. Solid-State Lett. 8 (2005) A603.
- [3] K. Shibata et al.: JPS Conf. Proc. 8 (2015) 036022.
- [4] K. Mori et al., Chem. Phys. Lett. 584 (2013) 113.



**Fig. 1.** Temperature dependence of S(Q, E) for  ${}^{7}\text{Li}_{7}P_{3}S_{11}$  crystal at  $Q = 0.32 \text{ Å}^{-1}$ : (a) 467 K, (b) 513 K, and (c) 552 K. The gray area indicates the quasielastic neutron scattering (QENS) component for the motion of Li ions.