実験報告書様式(一般利用課題・成果公開利用)

MLF Experimental Report	提出日 Date of Report		
課題番号 Project No.	装置責任者 Name of responsible person		
2015A0192	Miyake Yasuhiro		
実験課題名 Title of experiment	装置名 Name of Instrument/(BL No.)		
Investigation of chemical environmental effect in muonic atom	MUSE D1		
formation for low pressure gaseous molecules containing sulfur	実施日 Date of Experiment		
atoms	2016/4/25-26		
実験責任者名 Name of principal investigator			
Kazuhiko Ninomiya			
所属 Affiliation			
Osaka University			

試料、実験方法、利用の結果得られた主なデータ、考察、結論等を、記述して下さい。(適宜、図表添付のこと) Please report your samples, experimental method and results, discussion and conclusions. Please add figures and tables for better explanation.

tabi	es foi	better ex	pianation.						
1. 1	式料	Name of s	sample(s)	and chemical	formula.	or compositions	including physic	cal form.	

H2 gas

COS gas

CS2 gas

2. 実験方法及び結果(実験がうまくいかなかった場合、その理由を記述してください。)

Experimental method and results. If you failed to conduct experiment as planned, please describe reasons.

When a negatively charged muon is captured by an atom, muonic atom is formed. There are two process of muonic atom formation, one is a negative muon is directly captured by an atom (direct capture), and the other is muonic hydrogen atom is firstly formed and then the heavier nucleus takes the negative muon from the muonic hydrogen atom(muon transfer). The muon capture process is significantly affected by the chemical environment of the atom (chemical effect). In this study, we focused on muon capture probability of each atom and initial quantum state of the captured muon for COS and CS2 molecules through muonic atom formation by muon direct capture and muon transfer processes.

2. 実験方法及び結果(つづき) Experimental method and results (continued)

Figure 1 shows schematic view of experimental system. We irradiated muons in the low pressure ($^{\sim}$ 0.2 bar) gas samples and measured muonic X-rays by high-purity germanium detectors. The obtained muonic X-ray spectrum for COS sample is shown in Figure 2. From muonic X-ray intensities, we determined muon capture ratios as A(S/C) = 2.46+-0.12 for COS and A(S/C) = 3.35+-0.09 for CS2. Muon capture ratios were clearly changed by the molecular structure, and we are now starting discussion of chemical effect in muon capture process using molecular orbital calcurations.

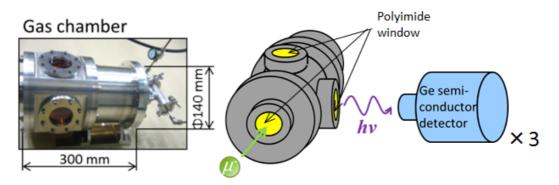


Figure 1. Schematic view of the experimental system.

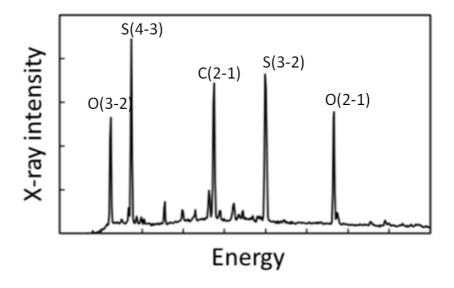


Figure 2. Muonic X-ray spectrum for COS sample.