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 MLF Experimental Report	提出日 Date of Report
課題番号 Project No. 2014B0063 実験課題名 Title of experiment Crystal structure of dielectric oxynitride perovskite: anionic distribution effect & its industrial application 実験責任者名 Name of principal investigator Yuji Masubuchi 所属 Affiliation Hokkaido University	装置責任者 Name of responsible person Shuki Torii 装置名 Name of Instrument/(BL No.) SuperHRPD / BL-08 実施日 Date of Experiment 2014/12/15-2014/12/16

試料、実験方法、利用の結果得られた主なデータ、考察、結論等を、記述して下さい。(適宜、図表添付のこと)
 Please report your samples, experimental method and results, discussion and conclusions. Please add figures and tables for better explanation.

<p>1. 試料 Name of sample(s) and chemical formula, or compositions including physical form.</p> <p>Name of sample: oxynitride perovskite Chemical formulae: 1. LaTiO_2N (powder), 2. $\text{La}_{0.8}\text{Sr}_{0.2}\text{TiO}_{2.2}\text{N}_{0.8}$ (powder), 3. SrTaO_2N (powder), 4. $\text{SrTaO}_2\text{N}_{0.9}$ (powder), 5. $\text{Ga}_{0.75}\text{Zn}_{0.25}\text{N}_{0.75}\text{O}_{0.25}$ (powder)</p>
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<p>2. 実験方法及び結果 (実験がうまくいかなかった場合、その理由を記述してください。) Experimental method and results. If you failed to conduct experiment as planned, please describe reasons.</p> <p>Experimental method: The oxynitride samples were prepared by nitridation of the oxide precursors under ammonia flow at 950 °C for 20 – 100 h. In order to investigate an effect of the nitrogen release on the crystal structure, SrTaO_2N was heat-treated at 1500 °C under 0.2 MPa of N_2 flow for 3 h. Crystalline phases were confirmed by powder XRD. Chemical compositions of the oxynitrides were analyzed by using both XRF and oxygen/nitrogen analyzer. Neutron diffraction patterns for the oxynitride samples were measured at room temperature by SuperHRPD. In the measurements, the powders were loaded in a vanadium can and then mounted in the apparatus. The computer program Z-Rietveld was used for the structural refinement.</p> <p>Results: Crystal structure of the LaTiO_2N perovskite has been reported in both space groups of orthorhombic Imma and triclinic I-1 (P-1) [1,2]. In the present study, diffraction line splitting observed by using the high resolution powder diffractometer is fitted well by the triclinic cell with space group of I-1.</p>
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2. 実験方法及び結果(つづき) Experimental method and results (continued)

Figure 1 shows a Rietveld refinement profile for LaTiO₂N by using the structural model with I-1. The refined structural parameters are listed in Table 1. There are three kinds of anionic sites and much amount of nitrogen was observed in O/N2 site. The other two sites were predominantly occupied by oxide ions. Preferred distribution of oxide and nitride ions may be related to the dielectric property of LaTiO₂N as suggested in similar oxynitride perovskite, SrTaO₂N [3]. We also analyzed the crystal structure of La_{0.8}Sr_{0.2}TiO_{2.2}N_{0.8} to investigate the relationship between the O/N ratio and the anionic distribution. The preferred anionic distribution of the oxynitride perovskite resulting from the non-stoichiometric composition has been investigated.

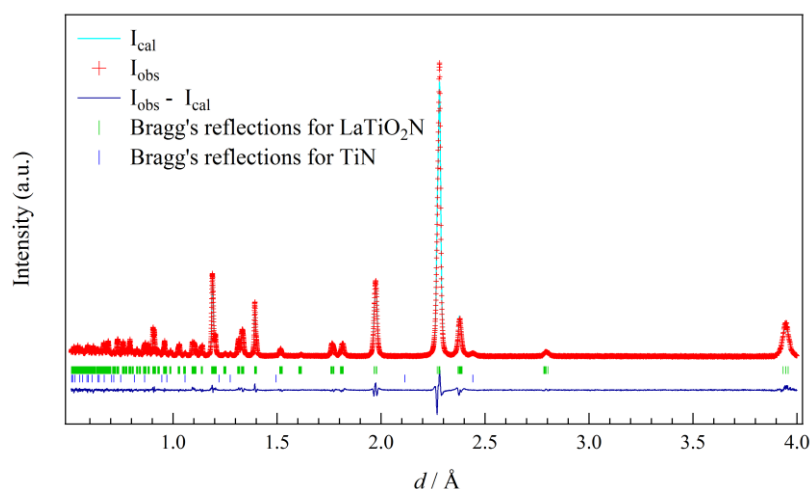


Fig. 1 Rietveld refinement profile for the high resolution neutron powder diffraction data of LaTiO₂N (S.G.; I-1).

Table 1 Structural parameters of the LaTiO₂N refined from the high resolution neutron powder diffraction data. Space group: I-1, ($a = 0.56055(1)$ nm, $b = 0.78685(1)$ nm, $c = 0.55742(1)$ nm, $\alpha = 90.205(2)^\circ$, $\beta = 90.164(2)^\circ$, $\gamma = 89.971(4)^\circ$, $R_{wp} = 5.88\%$, $R_e = 2.83\%$, $S^2 = 4.32$).

Atom	site	x	y	z	$B / \times 10^{-2} \text{ nm}^2$	Site occupancy
La	4 <i>i</i>	0.5031(5)	0.2486(5)	0.9993(5)	0.054(6)	1
Ti1	2 <i>a</i>	0	0	0	0.22(1)	1
Ti2	2 <i>f</i>	0	0.5	0	0.22(1)	1
O/N1	4 <i>i</i>	0.4405(6)	0.7482(6)	0.5097(6)	0.41(1)	0.776(4)/0.224
O/N2	4 <i>i</i>	0.2631(7)	0.0325(7)	0.2460(7)	1.05(3)	0.411(5)/0.589
O/N3	4 <i>i</i>	0.2619(8)	0.4681(8)	0.7598(8)	0.36(3)	0.814(5)/0.186

References

- [1] M. Yashima, M. Saito, H. Nakano, T. Takata, K. Ogisu, K. Domen, *Chem. Comm.*, **46** (2010) 4704-4706.
- [2] D. Logvinovich, L. Bocher, D. Sheptyakov, R. Figi, S. G. Ebbinghaus, R. Aguiar, M. H. Aguirre, A. Reller, A. Weidenkaff, *Solid State Sci.*, **11** (2009) 1513-1519.
- [3] Y.-R. Zhang, T. Motohashi, Y. Masubuchi, S. Kikkawa, *J. Ceram. Soc. Jpn.*, **119** (2011) 581-586.