


|   |   |
|---|---|
|  <b>MLF Experimental Report</b>  | 提出日 Date of Report<br>July 24, 2012   |
| 課題番号 Project No.<br>2012A0089<br>実験課題名 Title of experiment<br>Static/dynamic behavior in MgH <sub>2</sub> at high temperatures<br>実験責任者名 Name of principal investigator<br>Jun Sugiyama<br>所属 Affiliation<br>Toyota Central Research and Development Laboratories, Inc. | 装置責任者 Name of responsible person<br>Yasuhiro Miyake<br>装置名 Name of Instrument/(BL No.)<br>D1<br>実施日 Date of Experiment<br>June 06, 2012 – June 10, 2012 |

試料、実験方法、利用の結果得られた主なデータ、考察、結論等を、記述して下さい。(適宜、図表添付のこと)  
 Please report your samples, experimental method and results, discussion and conclusions. Please add figures and tables for better explanation.

|   |
|---|
| 1. 試料 Name of sample(s) and chemical formula, or compositions including physical form.  |
| <p>Two powder samples (MgH<sub>2</sub> and milled MgH<sub>2</sub>), which were prepared at Toyota CRDL, were packed into a gold O-ring sealed titanium cell in an Ar-filled glove box. Then, the Ti cell was mounted on a graphite sample holder in order to measure wTF-, ZF- and wLF-<math>\mu</math>SR spectrum at temperatures from 300 to 650 K under high vacuum. In order to avoid an unexpected explosion at high T due to the desorption reaction, we made small holes on the window— i.e. a Ti foil with 50 <math>\mu</math>m thickness, of the Ti cell just before the measurements.</p> |

|  |
|--|
| 2. 実験方法及び結果 (実験がうまくいかなかった場合、その理由を記述してください。)  |
| Experimental method and results. If you failed to conduct experiment as planned, please describe reasons.  |
| <p>Following upon the <math>\mu</math>SR measurements on a hydrogen storage material, MgH<sub>2</sub> below 500 K, we have measured wTF-, ZF-, and wLF-<math>\mu</math>SR spectra for an “as prepared MgH<sub>2</sub>” and “milled MgH<sub>2</sub>” sample until 750 K, in order to study the change in <math>\mu</math>SR parameters by the H-desorption reaction. Particularly, although there are no crucial changes in crystal structure between the “as prepared MgH<sub>2</sub>” sample and “milled MgH<sub>2</sub>” sample, the hydrogen desorption temperature (<math>T_d</math>) for the latter sample decreases by about 50 K compared with <math>T_d</math> for the former sample [1].</p> <p>Since the hydrogen desorption reaction of MgH<sub>2</sub> is an endothermic reaction, it was very difficult to control the sample temperature around <math>T_d</math>. In particular, the infrared lamp used as a heater of the present oven was found to respond too fast to stabilize the temperature around <math>T_d</math>. Therefore, for the milled sample, we obtained reliable data only below 570 K, whereas <math>T_d = 620</math> K. Figures 1(a) and 1(b) show the temperature dependences of the relaxation rate (<math>\lambda_{TF}</math>) of the wTF-spectrum, which corresponds to a spin-spin relaxation rate, i.e., the field distribution width (<math>\Delta</math>) due to <sup>1</sup>H nuclear field, and the field fluctuation rate (<math>\nu</math>) for the two samples. For the as prepared sample, both <math>\lambda_{TF}</math> and <math>\nu</math> are almost temperature independent up to 630 K, and then become very small, eventually 0 above <math>T_d</math>. This means</p> |

## 2. 実験方法及び結果(つづき) Experimental method and results (continued)

that the H-desorption reaction abruptly occurs and completes at  $T_d$ . In fact, at temperatures above  $T_d$ , the wTF-spectrum exhibited a non-relaxing oscillatory signal due to the applied wTF. Furthermore, the wTF-spectrum obtained in cooling mode from 750 K did not change even below  $T_d$ , suggesting that all the hydrogen atoms were removed from the sample.

On the other hand, for the milled sample,  $\lambda_{\text{TF}}$  gradually decreases with temperature above 470 K, while  $\nu$  apparently increases with temperature also above 470 K. This suggests that the H-desorption reaction starts to occur far below  $T_d$  (=620 K) for the milled sample. Moreover, the increase in  $\nu$  implies rapid diffusion of  $\text{H}^+$  and/or  $\mu^+$  in the sample, which is proposed to be a significant factor to decrease  $T_d$  for the other hydrogen storage materials, such as  $\text{LiAlH}_4$  [2] and  $M(\text{BH}_4)_n$  [2].

However, in order to further elucidate the mechanism on the decrease in  $T_d$  by milling and/or by adding a small amount of  $\text{Nb}_2\text{O}_5$ , we need more systematic measurements at high temperatures, particularly below the vicinity of  $T_d$  using a more reliable temperature control system.

### REFERENCES

- [1] N. Hanada, et al., J. Alloys and Compounds **446-447**, 67 (2007).
- [2] R. Kadono, et al., Phys. Rev. Lett. **100**, 026401 (2008).
- [3] J. Sugiyama, et al., Phys. Rev. B **81**, 092103 (2010).

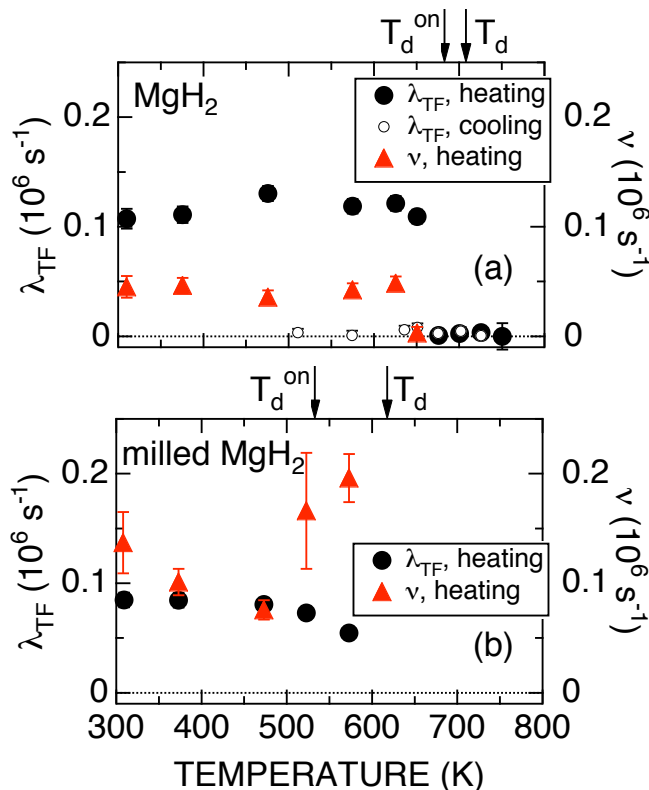


Fig. 1: The temperature dependences of the wTF relaxation rate ( $\lambda_{\text{TF}}$ ) and the field fluctuation rate ( $\nu$ ) for (a) the as prepared  $\text{MgH}_2$  and (b) milled  $\text{MgH}_2$  sample. The data of  $\lambda_{\text{TF}}$  were obtained by fitting the wTF-spectrum using an exponentially relaxing cosine oscillation function, and that of  $\nu$  were obtained by fitting the ZF- and wLF-spectra using a dynamic Kubo-Toyabe function. Here, wTF was 20 Oe and wLF was 10 Oe.